

OTC Website Find Technologies Contact Us

Request Information Permalink

# Magnesium Enhanced Reactivity of High Energy Composites

Tech ID: 33257 / UC Case 2023-971-0

# **FULL DESCRIPTION**

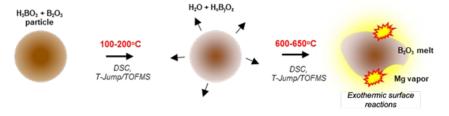
#### **Background**

Boron (B) is regarded as a premier candidate fuel in high-energy composites due to its higher reaction enthalpies. However, boron suffers from sluggish oxidation and energy release kinetics as a result of its low melting oxide shell. Post-melting, the non-volatile liquid oxide layer acts as a diffusion barrier to the oxidizing species and restricts their access to the boron core, thereby inhibiting oxidation and energy release.

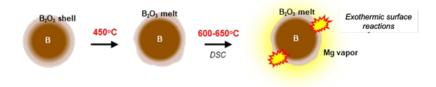
## Technology

Prof. Zachariah and his team have developed an innovative magnesium/boron (Mg/B) based composite that offers a thermodynamically and kinetically viable source of highly reactive gas-phase Mg that acts as an etchant for the oxide shell of boron. The developed composite creates a pristine and accessible fuel surface for the reaction. Further reactions are facilitated by the thinning of the oxide shell. Together, these serve to enhance the combustion of boron.

# a. Mg/B<sub>2</sub>O<sub>3</sub> reaction mechanism:



b. Mg vapor-assisted surface disruption of oxide shell of B NPs:



Schematic illustration of the reactivity enhancement of Boron

# CONTACT

Venkata S. Krishnamurty
venkata.krishnamurty@ucr.edu
tel:

### OTHER INFORMATION

#### **KEYWORDS**

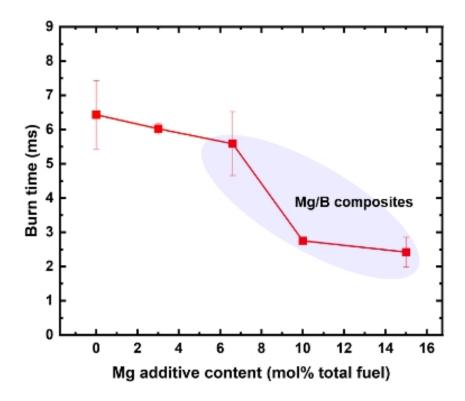
energetic composites, thermites,
propellants, pyrotechnics, solid fuels

#### **CATEGORIZED AS**

- **▶** Energy
  - ▶ Other
- ► Materials & Chemicals
  - ▶ Composites
  - ▶ Nanomaterials

## **RELATED CASES**

2023-971-0



Graph showing the burn characteristics of B/CuO nanoenergetic composites. Mg/B/CuO composites show significant enhancement in burn times and pressurization over B/CuO composites.

# **ADVANTAGES**

The significant benefits and aspects of this Mg/B-based composites relative to B-based composites are:

- A 6-fold enhancement in pressurization rates.
- ► A 30% increase in peak pressures.
- ▶ Shortened burn time from 6.5 milliseconds (ms) to ~2 ms a 60% reduction.
- Synergistic effect of the Mg and B fuels responsible for the augmentation of reactivity.
- ▶ Potential for similar augmentation for high-energy composites based on nanoscale metals and metalloids such as Aluminum, Titanium, Silicon, etc.

# **SUGGESTED USES**

# **USER DEFINED 1**

- Please review all inventions by Prof. Zachariah and his team at UCR.
- ▶ Please read recent press coverage of Prof. Zachariah, at UCR.
- ▶ Please visit Prof. Zachariah's research group website to learn more about their research.

# **RELATED MATERIALS**

▶ Magnesium-Enhanced Reactivity of Boron Particles; Role of Mg/B2O3 Exothermic Surface Reactions

# **PATENT STATUS**

Patent Pending

# **RELATED TECHNOLOGIES**

- ▶ One-Step Synthesis of Aligned Nanoparticles With High Purity
- ▶ Unzipping Polymers For Enhanced Energy Release

University of California, Riverside

Office of Technology Commercialization

200 University Office Building,

Riverside,CA 92521

otc@ucr.edu

research.ucr.edu/

Terms of use | Privacy Notice | © 2023, The Regents of the University of California