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Platinum Oxide Nanoparticles For Electrocheical Hydrogen Evolution Influence Of Platinum Valence State

Tech ID: 33005 / UC Case 2020-259-0

BACKGROUND

Platinum remains a leading choice of catalyst for the Hydrogen Evolution reaction (HER) but because of its high cost and low natural abundance, it is critical to optimize its use. HER catalysts with reduced amounts of Pt would be of high value.

TECHNOLOGY DESCRIPTION

Platinum oxide nanoparticles of approximately 2 nm in diameter are deposited on carbon nitride (C₃N₄) nanosheets by thermal refluxing of C₃N₄ and PtCl₂ or PtCl₄ in water. These nanoparticles exhibit apparent electrocatalytic activity toward the hydrogen evolution reaction (HER) in acid. Interestingly, the HER activity increases with increasing Pt⁴⁺ concentration in the nanoparticles, and the optimized catalyst even outperforms commercial Pt/C, exhibiting an overpotential of only -7.7 mV to reach the current density of 10 mA cm⁻² and a Tafel slope of -26.3 mV dec⁻¹.

Description unavailable

APPLICATIONS

Hydrogen evolution reaction in acidic conditions

Clean hydrogen production by electrolysis

ADVANTAGES

Maximizing Pt⁴⁺ ions relative to Pt²⁺ ions in nanoparticles results in a very high efficiency HER catalyst while minimizing the amount of platinum used.

INTELLECTUAL PROPERTY INFORMATION

Country	Type	Number	Dated	Case
United States Of America	Issued Patent	11,767,602	09/26/2023	2020-259
United States Of America	Published Application	20230383425	11/30/2023	2020-259

RELATED MATERIALS

▶ [Platinum Oxide Nanoparticles for Electrochemical Hydrogen Evolution: Influence of Platinum Valence State - 10/28/2019](#)

CONTACT

Jeff M. Jackson
jjackso6@ucsc.edu
 tel: .



INVENTORS

- ▶ Chen, Shaowei
- ▶ Nichols, Forest

OTHER INFORMATION

KEYWORDS

Hydrogen Evolution Reaction,
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 Catalysts, Hydrogen Production by
 Electrolysis, Clean Hydrogen
 Production, Platinum Catalyst,
 Hydrogen Evolution Reaction Acidic
 Conditions, Platinum Nanoparticles

CATEGORIZED AS

- ▶ [Energy](#)
- ▶ [Hydrogen](#)

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ADDITIONAL TECHNOLOGIES BY THESE INVENTORS

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University of California, Santa Cruz

Industry Alliances & Technology Commercialization

Kerr 413 / IATC,

Santa Cruz, CA 95064

Tel: 831.459.5415

innovation@ucsc.edu

officeofresearch.ucsc.edu/

Fax: 831.459.1658

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